

**AMENDMENTS TO THE CLAIMS**

Please amend the claims as follows:

Claim 1 (Currently Amended):           A method for forming a capacitor in a semiconductor device, comprising:

forming a lower electrode constituted with a silicon ~~oxide~~ layer on a semiconductor substrate by a predetermined process on which a predetermined process has been completed;

forming a uniform silicon oxide layer on the lower electrode by performing an atomic layer deposition (ALD) process;

forming an aluminum oxide ( $\text{Al}_2\text{O}_3$ ) film on the silicon oxide layer by using an ALD method, wherein the method for forming the uniform silicon oxide layer and the aluminum oxide film ~~to reduce~~ incubation time required for the formation of the  $\text{Al}_2\text{O}_3$  film on the silicon oxide layer and to remove prevents metallic clusters from formed forming at an interface between the  $\text{Al}_2\text{O}_3$  film and the silicon oxide layer; ~~and~~ crystallizing the  $\text{Al}_2\text{O}_3$  film by carrying out a heat treatment process, and forming an upper electrode on the  $\text{Al}_2\text{O}_3$  film, wherein the upper electrode includes multi-layers of a polysilicon layer and a metal layer.

Claim 2 (Canceled)

Claim 3 (Original):     The method as recited in claim 1, wherein the silicon oxide layer is formed by using an in-situ method or an ex-situ method.

Claim 4 (Original):     The method as recited in claim 1, wherein a silicon source selected from a group consisting of  $\text{SiCl}_4$ , DCS and HCD and a reaction source selected from a group consisting of  $\text{H}_2\text{O}$ ,  $\text{O}_3$  and  $\text{H}_2\text{O}_2$  are used to form the silicon oxide layer during the ALD process.

Claim 5 (Original):     The method as recited in claim 4, wherein a pyridine acting as a catalyst is used when the silicon source and the reaction source are supplied during the ALD process.

Claim 6 (Original): The method as recited in claim 4, wherein each of a supply time and a purge time for the silicon source and the reaction source is less than 10 seconds respectively.

Claim 7 (Previously Presented): The method as recited in claim 1, wherein the silicon oxide layer is formed at a low temperature less than about 200 °C.

Claim 8 (Original): The method as recited in claim 7, wherein a thickness of the silicon oxide layer is less than about 10 Å.

Claim 9 (Canceled)

Claim 10 (Previously Presented): The method as recited in claim 1, wherein  $\text{Al}(\text{CH}_3)_3$ , which is trimethylaluminum (TMA), is used as an aluminum source, and one of  $\text{H}_2\text{O}$ ,  $\text{O}_3$  and  $\text{H}_2\text{O}_2$  is used as a reaction source during the ALD process to form the  $\text{Al}_2\text{O}_3$  film.

Claim 11 (Previously Presented): The method as recited in claim 10, wherein a plasma is used as an energy source during the ALD process to form the  $\text{Al}_2\text{O}_3$  film.

Claim 12 (Previously Presented): The method as recited in claim 11, wherein the ALD process to form the  $\text{Al}_2\text{O}_3$  film is carried out at a room temperature or at a temperature of about 500 °C.

Claim 13 (Previously Presented): The method as recited in claim 1, wherein a thickness of the  $\text{Al}_2\text{O}_3$  film is less than about 100 Å.

Claim 14 (Original): The method as recited in claim 1, wherein the heat treatment process is carried out at a temperature greater than 600 °C and in an  $\text{N}_2$  or  $\text{O}_2$  ambient.

Claim 15 (Original): The method as recited in claim 14, wherein the heat treatment process is carried out by using a furnace annealing process or a rapid thermal process (RTP).

Claim 16 (Original): The method as recited in claim 1, wherein an upper electrode constituted with a metal layer, a silicon layer or a metal layer/silicon layer is formed on an upper area of the crystallized  $\text{Al}_2\text{O}_3$  film.

Claim 17 (New): The method as recited in claim 1, wherein the silicon layer includes one of an undoped polysilicon layer, and a doped amorphous silicon layer.